

Yue-Qing Pu,\* Hai-Bo Wang,  
Jia-Hui Chen and Jin-Tang WangDepartment of Applied Chemistry, College of  
Science, Nanjing University of Technology,  
Xinmofan Road No. 5 Nanjing, Nanjing  
210009, People's Republic of ChinaCorrespondence e-mail:  
wanghaibo@njut.edu.cn

## Key indicators

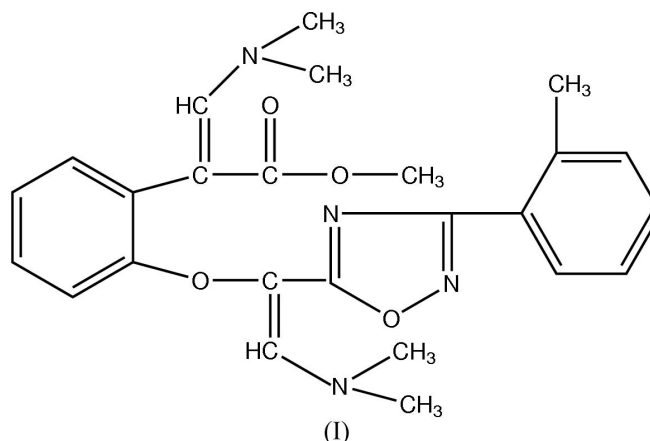
Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å  
 $R$  factor = 0.060  
 $wR$  factor = 0.193  
Data-to-parameter ratio = 15.7For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.Methyl 3-(dimethylamino)-2-(2-{2-(dimethylamino)-  
1-[3-(2-methylphenyl)-1,2,4-oxadiazol-5-yl]vinyl-  
oxy}phenyl)acrylateThe title compound,  $\text{C}_{25}\text{H}_{28}\text{N}_4\text{O}_4$ , was synthesized by the  
reaction of methyl 2-[[3-(2-methylphenyl)-1,2,4-oxadiazol-5-  
yl]methoxy]phenyl acetate and *N,N*-dimethylformamide  
dimethyl acetal. In the crystal structure, there are intra-  
molecular  $\text{C}-\text{H}\cdots\text{O}$  and  $\text{C}-\text{H}\cdots\text{N}$  hydrogen bonds and  
intermolecular  $\text{C}-\text{H}\cdots\pi$  interactions.

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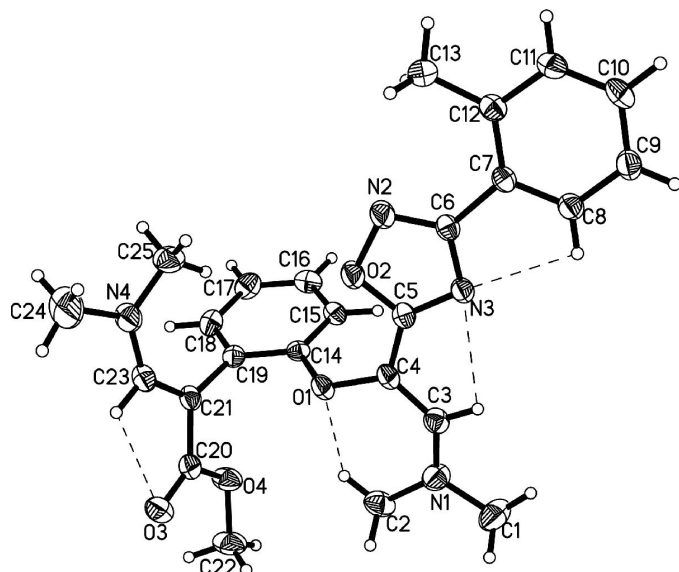
Online 23 April 2005

## Comment

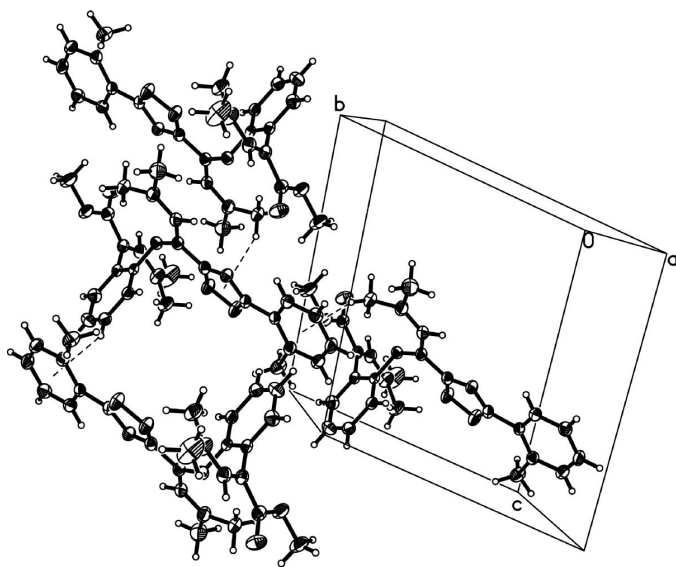
1,2,4-Oxadiazoles represent an important class of five-  
membered heterocycles. Some derivatives of 1,2,4-oxadiazoles  
have intrinsic analgesic (Terashita *et al.*, 2002), anti-inflamma-  
tory (Nicolaides *et al.*, 1998) and antipicornaviral (Romero,  
2001) properties and are efficient as agonists [*e.g.* forangi-  
tensin (Naka *et al.*, 1999) and adhesion agents (Juraszyk *et al.*,  
1997)] for different receptors. We report here the crystal  
structure of the title compound, (I).The molecular structure of (I) is shown in Fig. 1, where the  
dashed lines indicate intramolecular  $\text{C}-\text{H}\cdots\text{O}$  and  $\text{C}-  
\text{H}\cdots\text{N}$  hydrogen bonds. Selected bond lengths and angles are  
given in Table 1. There are also  $\text{C}-\text{H}\cdots\pi$  interactions in the  
crystal structure (Fig. 2). Full details of the hydrogen bonding  
are given in Table 2. The combination of both types of weak  
interactions generates a three-dimensional network.

## Experimental

Methyl 2-[[3-(2-methylphenyl)-1,2,4-oxadiazol-5-yl]methoxy]phenyl  
acetate (14 mmol) was dissolved in dimethylformamide (20 ml) and  
*N,N*-dimethylformamide dimethyl acetal (8 ml) was added in one  
portion. The resulting mixture was refluxed for 6 h and then  
concentrated under reduced pressure to afford crude compound (I).  
Pure compound (I) was obtained by crystallization from ethyl acetate  
(15 ml) and petroleum ether (7.5 ml). Crystals of (I) suitable for X-ray



**Figure 1**  
A view of the molecular structure of (I). Dashed lines indicate intramolecular C—H...O and C—H...N hydrogen bonds. Displacement ellipsoids are drawn at the 30% probability level.



**Figure 2**  
The C—H... $\pi$  interactions in (I), shown as dashed lines.

diffraction were obtained by slow evaporation of an ethanol solution. Spectroscopic analysis:  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ,  $\delta$ , p.p.m.): 7.86–7.88 (*m*, 1H), 7.75 (*m*, 1H), 7.35–7.37 (*m*, 1H), 7.30–32 (*m*, 2H), 7.28–7.30 (*m*, 1H), 7.23–7.25 (*m*, 2H), 7.01 (*m*, 1H), 6.95–6.97 (*m*, 1H), 3.61 (*m*, 3H), 3.06 (*s*, 6H), 2.89 (*m*, 6H), 2.59 (*s*, 3H).

#### Crystal data

$\text{C}_{25}\text{H}_{28}\text{N}_4\text{O}_4$   
 $M_r = 448.51$   
 Triclinic,  $P\bar{1}$   
 $a = 10.250$  (2) Å  
 $b = 10.691$  (2) Å  
 $c = 12.263$  (3) Å  
 $\alpha = 105.46$  (3)°  
 $\beta = 110.74$  (3)°  
 $\gamma = 92.30$  (3)°  
 $V = 1197.7$  (6) Å<sup>3</sup>  
 $Z = 2$   
 $D_x = 1.244$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 Cell parameters from 25 reflections  
 $\theta = 10$ –13°  
 $\mu = 0.09$  mm<sup>-1</sup>  
 $T = 293$  (2) K  
 Block, colourless  
 $0.4 \times 0.3 \times 0.3$  mm

#### Data collection

Enraf–Nonius CAD-4  
 diffractometer  
 $\omega/2\theta$  scans  
 Absorption correction: none  
 4945 measured reflections  
 4671 independent reflections  
 2950 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.024$

$\theta_{\text{max}} = 26.0^\circ$   
 $h = 0 \rightarrow 12$   
 $k = -12 \rightarrow 12$   
 $l = -14 \rightarrow 14$   
 3 standard reflections  
 every 200 reflections  
 intensity decay: none

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.060$   
 $wR(F^2) = 0.193$   
 $S = 1.05$   
 4671 reflections  
 298 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.1P)^2 + 0.19P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.32 \text{ e } \text{Å}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.31 \text{ e } \text{Å}^{-3}$

**Table 1**

Selected geometric parameters (Å, °).

O1—C14	1.391 (3)	N3—C6	1.381 (3)
O1—C4	1.400 (3)	N4—C23	1.347 (4)
O2—C5	1.343 (3)	N4—C25	1.444 (4)
O2—N2	1.429 (3)	N4—C24	1.447 (4)
O3—C20	1.214 (3)	C3—C4	1.354 (4)
O4—C20	1.357 (3)	C4—C5	1.435 (4)
O4—C22	1.444 (3)	C6—C7	1.487 (3)
N1—C3	1.341 (3)	C12—C13	1.502 (4)
N1—C1	1.450 (4)	C19—C21	1.483 (4)
N1—C2	1.457 (4)	C20—C21	1.458 (4)
N2—C6	1.303 (3)	C21—C23	1.360 (4)
N3—C5	1.296 (3)		
C14—O1—C4	117.52 (18)	N2—C6—N3	114.6 (2)
C5—O2—N2	106.35 (19)	N2—C6—C7	124.5 (2)
C20—O4—C22	116.2 (2)	N3—C6—C7	120.9 (2)
C3—N1—C1	120.5 (3)	C8—C7—C6	117.0 (2)
C3—N1—C2	123.6 (3)	C12—C7—C6	123.0 (2)
C1—N1—C2	115.9 (3)	C11—C12—C13	118.3 (3)
C6—N2—O2	102.9 (2)	C7—C12—C13	124.5 (2)
C5—N3—C6	103.1 (2)	C15—C14—O1	122.8 (2)
C23—N4—C25	123.7 (2)	C19—C14—O1	115.6 (2)
C23—N4—C24	120.1 (3)	C18—C19—C21	120.8 (2)
C25—N4—C24	116.2 (3)	C14—C19—C21	122.3 (2)
N1—C3—C4	132.3 (3)	O3—C20—O4	121.2 (3)
C3—C4—O1	123.9 (2)	O3—C20—C21	127.3 (3)
C3—C4—C5	119.4 (2)	O4—C20—C21	111.4 (2)
O1—C4—C5	116.6 (2)	C23—C21—C20	114.3 (2)
N3—C5—O2	113.0 (2)	C23—C21—C19	126.1 (2)
N3—C5—C4	128.8 (2)	C20—C21—C19	119.4 (2)
O2—C5—C4	118.2 (2)	N4—C23—C21	132.5 (3)

**Table 2**

Hydrogen-bond geometry (Å, °).

$D\text{—H}\cdots A$	$D\text{—H}$	$\text{H}\cdots A$	$D\cdots A$	$D\text{—H}\cdots A$
C2—H2B...O1	0.96	2.19	2.963 (4)	136
C3—H3B...N3	0.93	2.54	2.922 (3)	105
C8—H8A...N3	0.93	2.48	2.839 (3)	103
C23—H23A...O3	0.93	2.35	2.771 (4)	107
C2—H2C...Cg1 <sup>i</sup>	0.96	2.78	3.667 (4)	153
C2—H2D...Cg2 <sup>ii</sup>	0.96	2.75	3.674 (3)	161
C16—H16A...Cg2 <sup>iii</sup>	0.93	2.71	3.531 (4)	148

Symmetry codes: (i)  $-x, -y + 1, -z + 1$ ; (ii)  $x, y + 1, z$ ; (iii)  $-x + 1, -y + 1, -z + 2$ .  
 Notes: Cg1 is the centroid of ring C5/O2/N2/C6/N3 and Cg2 is the centroid of ring C7—C12.

All H atoms were placed geometrically, with C—H distances in the range 0.93–0.96 Å, and included in the refinement in a riding-model approximation, with  $U_{\text{iso}}(\text{H}) = 1.2$  or  $1.5U_{\text{eq}}(\text{C})$ .

Data collection: *CAD-4 Software* (Enraf-Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXL97*.

## References

Enraf-Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf-Nonius, Delft, The Netherlands.

Harms, K. & Wocadlo, S. (1995). *XCAD4*. University of Marburg, Germany.  
Jurazyk, H., Gante, J., Wurziger, H., Bernotat-Danielowski, S. & Melzer, G. (1997). PCT Int. Appl. 9744333.  
Naka, T. & Kubo, K (1999). *Curr. Pharm. Des.* **5**, 453–472.  
Nicolaidis, D. N., Fylaktakidou, K. C., Litinas, K. E. & Hadjipavlou-Litina, D. (1998). *Eur. J. Med. Chem.* **33**, 715–724.  
Romero, J. R. (2001). *Expert Opin. Invest. Drugs*, **10**, 369–379.  
Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.  
Siemens (1996). *SHELXTL*. Version 5.06. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.  
Terashita, Z., Naruo, K. & Morimoto, S. (2002). PCT Int. Appl. WO, 0260439.